



## **Program FYSICA 2019**

Friday, April 5

Amsterdam Science Park

[www.fysica.nl](http://www.fysica.nl)

Parallel Session title: **Power2Fuel**

Abstract: The Paris agreement (April 2016) leaves no doubt that for both energy plants as well as petro-chemistry, carbon dioxide emissions should be completely banned in 2050. This goal represents a huge challenge for both fundamental science and technology, but also for politics in an international context. For Northern Europe, the main sources for electrical energy will be solar and wind energy. These sources are both discontinuous in their supply, but will have to be matched with a continuous demand. Especially for slow time scale variations (seasonal time scale) this cannot be solved using batteries. This leads to the need for the production of carbon based energy-dense chemicals using renewable electrical energy, using atmospheric CO<sub>2</sub> on a large scale. To be able to reduce atmospheric CO<sub>2</sub> in the nearer future, carbon capture and storage (CCS) is topical. In this Power2Fuel session, the latest developments concerning the above mentioned fields are presented

Convener: Lodewijk Arntzen (HHS)

Speakers (in order of appearance)

Title: **Towards the sustainable production of carbon based chemicals and fuels from renewable resources**

Wilson Smith (TUD)

Abstract: Electrocatalytic CO<sub>2</sub> reduction has the dual-promise of neutralizing carbon emissions in the near future, while providing a long-term pathway to create energy-dense chemicals and fuels from atmospheric CO<sub>2</sub>. The field has advanced immensely in recent years, taking significant strides towards commercial realization. While catalyst innovations have played a pivotal role in increasing the product selectivity and activity of both C1 and C2 products, slowing advancements indicate that electrocatalytic performance may be approaching a hard cap. Meanwhile, innovations at the systems level have resulted in the intensification of CO<sub>2</sub> reduction processes to industrially-relevant current densities by using pressurized electrolytes, gas-diffusion electrodes and membrane-electrode assemblies to provide ample CO<sub>2</sub> to the catalyst. The immediate gains in performance metrics offered by operating under excess CO<sub>2</sub> conditions goes beyond a reduction of system losses and high

current densities, however, with even simple catalysts outperforming many of their H-cell counterparts. Using recent literature as a guidepost, this talk will focus on some of the underlying reasons for the observed changes in catalytic activity, and proposes that further advances can be made by shifting additional efforts in catalyst discovery and fundamental studies to system-integrated testing platforms.

**Title: CO<sub>2</sub> capture from flue gas and the air: a short review**

Hans Geerlings (TUD)

**Abstract:** CO<sub>2</sub> capture from industrial gases like flue gases is a well-established technology that is available for deployment at large scale. Capture of CO<sub>2</sub> from the air, which is a key technology in envisaged solar fuel synthesis schemes, has not reached that state of development yet. In this overview, the physics of both families of capture technologies will be discussed with the emphasis on differences and similarities. It will be made clear that CO<sub>2</sub> capture from the air requires dedicated development effort both at a fundamental and technical development level.

**Title: Renewable energy driven chemistry for the production of fuels and chemicals**

Richard van de Sanden (DIFFER & TU/e)

**Abstract:** The next 20 years will show a fundamental change in the way energy is generated, stored, distributed, valued and consumed. Renewable energy generation by means of wind, from solar radiation through photovoltaics or concentrated solar power will continue to increase its share in the energy mix. Therefore economical and scalable methods for either the direct conversion of solar photons into chemicals and fuels, or conversion via the renewable electricity intermediary are urgently needed. Intermittency (due to e.g. day/night cycle), the regional variation of these energy sources, seasonal variation and penetration of the use of renewable energy into other sectors than the electricity sector (e.g. the chemical industry) requires means to store, transport and convert energy on a large scale. A promising option is the synthesis of chemicals and synthetic fuels (easily deployable within the present fossil fuels infrastructure) from raw feedstock using renewable energy. A truly circular economy requires that the raw materials are the thermodynamically most stable molecules such as water (H<sub>2</sub>O), carbon dioxide (CO<sub>2</sub>) and nitrogen (N<sub>2</sub>) to generate, e.g. hydrogen, hydrocarbons and ammonia. In this talk it will be highlighted how renewable energy driven chemistry can potentially combine compatibility with e.g. intermittency and localized production.

**Title: Geological storage of CO<sub>2</sub>: it's all about the physics**

Suzanne Hangx (UU)

**Abstract:** In order to curb global CO<sub>2</sub> emissions, carbon capture and storage (CCS) is considered to be one of the technologies able to reduce emissions on a short to medium time scale, with global storage capacities ranging from 100's to 1000's of gigatonnes. Suitable storage locations under

consideration are depleted oil/gas reservoirs and deep, saline aquifers. However, injecting CO<sub>2</sub> into the subsurface removes the natural system from its chemical and physical equilibrium. Fluid injection inevitably leads to a (poro)elastic response of the storage reservoir. In addition, high-pressure CO<sub>2</sub> injection will lead to thermal effects and acidification of the reservoir pore fluid. These in turn may impact the hydrological and mechanical behaviour of not just the reservoir, but also the overlying seal formation and any pre-existing faults within the storage complex. To ensure safe, long-term storage is therefore dependent upon maintaining containment integrity, i.e. keeping the stored CO<sub>2</sub> in the subsurface for > 10,000 yrs. Research has extensively focused on understanding this complex interplay between thermal, hydrological, chemical and mechanical processes. This has led to clear implications for site selection and assessing safety, but perhaps this knowledge can also form a basis for understanding future uses of the subsurface, such as energy storage or geothermal energy production.